## **Preparation of a New Polyoxometalate-based Nanoparticles**

Xiao Hong WANG<sup>1</sup>, Feng LI<sup>1</sup>, Jing Fu LIU<sup>1</sup>\*, M. T. POPE<sup>2</sup>

<sup>1</sup>Department of Chemistry, Northeast Normal University, Changchun 130024 <sup>2</sup>Department of Chemistry, Georgetown University, Washington DC 20057-1227, U. S. A

**Abstract:** Polyoxometalates (POMs)  $\alpha$ -K<sub>8</sub> H<sub>6</sub> [Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub>O<sub>77</sub>] (Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub>) loaded starch nanoparticles have been prepared and structurally characterized by elemental analyses, IR spectra and <sup>29</sup>Si spectroscopy. The particle size of Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub>/starch was estimated by a Transmission electron microscope (TEM) and a 1000HSA MALVIRN Zetasizer instrument. The result shows that the polyoxometalate retained the parent structure after encapsulated by starch microspheres.

Keywords: Polyoxometalate, nanoparticle, synthesis.

In recent years, considerable progress has been made to synthesize polyoxometalates (POMs) with nanosize<sup>1,2</sup>, but these methods were still limited in application of POMs used as drugs delivery<sup>3,4</sup>.

Such microspheres starch and liposome used as drug carrier for therapeutic applications are biocompatible, biodegradable and physically stable<sup>5,6</sup> and show some benefits including solubilization of poorly soluble drugs, local depot for the sustained release of topically effective drugs, or reduction of side effects or incompatibilities.

Currently, we are exploring the possibility of combining drug delivery technology and POMs syntheses to prepare new POMs complexes materials in order to solve the problem that most POMs compounds are not stable at physiological pH conditions. Here in, we report the preparation of starch loading POMs  $\alpha$ -K<sub>8</sub>H<sub>6</sub> [Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub>O<sub>77</sub>]<sup>7</sup> nanoparticles.

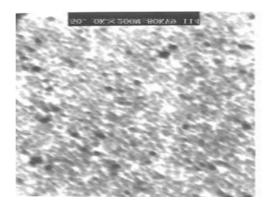
In the synthesis, starch microspheres have been prepared from low cost raw materials-soluble starch by reversed-phase microemulsion polymerization methods using phosphorous oxychloride as the linking agent<sup>8</sup>. The titanium polyoxotungstates  $Si_2W_{18}Ti_6$  loaded starch nanoparticles have been prepared by an enveloping method. 1.5 g of soluble starch was added to 10 mL of water to form a 15% mixture. The mixture was heated in boiling water-bath at a certain time until the mixture turned to transparent to form starch water aqueous. This aqueous solution was then cooled to room temperature, 200mg of  $\alpha$ -Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub><sup>8</sup> was added to it to form a water phase. This water phase was added drop-wise to an oil-phase (the oil phase contains 200 mL of C<sub>6</sub>H<sub>5</sub>CH<sub>3</sub>, 200 mL of CHCl<sub>3</sub> and a certain amount of surfactant Span-80) with stirring. It was kept stirring until

<sup>\*</sup> E-mail: sdjingfu@public.cc.jl.cn

a microemulsion was formed, the microemulsion was then treated by supersonic for 30 min in order to obtain smaller size nano-starch particles. To this microemulsion, 3 mL of POCl<sub>3</sub> was added and stirring was continued for another 1h. The microemulsion was left statically to obtain two layers. The water phase was separated and washed with acetone and ethanol to obtain the white solid (1g), and then the solid was washed with small quantities of water for four times in order to separate the unencapsulated  $\alpha$ -Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub>. The solid was dried at 60°C under vacuum to obtain the dry powder. The contents of  $\alpha$ -Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub> were analyzed by ICP-absorption spectroscopy.

From the elementary analytic results of the POMs/starch, the contents of every elements are W, 7.30; Si, 0.13; Ti, 0.64; P, 1.66%. It can be seen that during the POMs were encapsulated by starch microspheres, the ratio of W: Ti: Si is still 9:3:1 consistent with the encapsulation of  $\alpha$ -Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub>, indicating that the Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub> did not change during the reaction. The content of Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub> in starch microspheres was 12.0%, according to the amount of Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub> used in the experiment, the encapsulating effectiveness of starch microsphere to Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub> can be calculated to be 60%. The IR spectrum of the POMs/starch nanoparticles showed the absorption peaks at 993, 953, 896, 783 and 720 cm<sup>-1</sup>, which are characteristic asymmetric stretching vibrational peaks for heteropolyanions with (dimeric) Keggin structure of W-O<sub>d</sub>, W-O<sub>b</sub>-W, W-O<sub>c</sub>-W and Si-O<sub>a</sub>. The peak in 1158 cm<sup>-1</sup> shows a C-O-P vibration peak, which shows that during the starch reacting with POCl<sub>3</sub>, the cross-linking reaction occurred. The content of P atom in Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub>/starch microsphere (1.45%) also confirms this point.

The <sup>29</sup>Si MAS NMR spectrum of the POMs/starch nanoparticles shows one peak at -84.90 ppm as the same as that of pure  $\alpha$ - Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub> ( $\delta$ : -83.20 ppm). This shows that the primary  $\alpha$ - Si<sub>2</sub>W<sub>18</sub>Ti<sub>6</sub> structure is retained after encapsulation of the POMs.



#### Figure 1 The TEM of the POMs / starch

The transmission electron micrographs (TEM) (**Figure 1**) show that the POMs / starch forms relatively uniform nanometer particles of average diameter about 40-60 nm, and the shape of the nanoparticles is round and the particles did not agglomerated. The range of nanoparticles estimated by Zatesizer is from 100 nm to 20 nm, the size of the most nanoparticles is 40 nm.

# 716 Preparation of a New Polyoxometalate-based Nanoparticles

## Acknowledgments

This work was supported by the National Natural Science Foundation of China (No. 20171011) and the Science Foundation of Northeast Normal University.

### References

- 1. E. Coronado, C. J. Gomez-Gracia, J. Comments Inorg. Chem., 1995, 17, 255.
- 2. E. Coronado, P. Delhaes, J. R. Galan Mascaros, et al., Synth. Met., 1997, 85, 1647.
- 3. J. T. Rhule, C. L. Hill, D. A. Judd, Chem. Rev., 1998, 98, 327.
- 4. T. Yamase, H. Fujita, K. Fukushima, Inorg. Chim. Acta, 1988, 151, 15.
- 5. T. Murata, K. Akagi, M. Imamura, et al., Oncology, Reports, 1998, Vol. 5, 709.
- 6. M. Y. Levy, S. Benita, Int.J. Pharm., 1989, 54, 103.
- 7. J. F. Liu, L. Meng, W. Y. Zhao, B. L. Zhao, Acta Chimica Sinica, 1995, 53, 46.
- 8. L.Levier, D. E. Baker, Hosp. Pharm., 1993, 28, 1214.

Received 13 May, 2003